

Glue this side down into your ISN using only

4 dots of glue

"A dot is a lot!"



Fold along line

# CELL TONICITY

Using the key below and the information given, answer the questions.

**key:**

solute particle •

cell membrane - - - - -

cell wall        = = = =

in all solutions, the solvent is  $H_2O$

**Part I. Fill in the blanks:**

A \_\_\_\_\_ is a fluid in which a substance is dissolved.

A \_\_\_\_\_ is a substance dissolved in a solvent.

A \_\_\_\_\_ is a combination of solute and solvent.

The process by which  $H_2O$  diffuses across a membrane is called \_\_\_\_\_.

**Part II. Look at the solutions illustrated below and fill in the blanks.**

1. **Solution B** is \_\_\_\_\_ to **solution A**. This is because **solution B** has a greater concentration of \_\_\_\_\_ in it than does **solution A**. **Solution C** has no solutes dissolved in it, therefore it is \_\_\_\_\_ to both **Solutions A** and **B**.

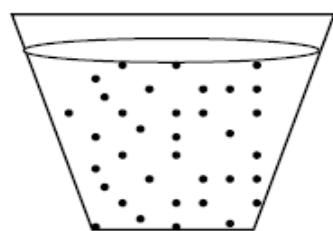
2. As the relative concentration of **solutes** in two solutions increases, of necessity the relative concentration of **water** in the same two solutions \_\_\_\_\_. **Solution A** has a lower concentration of \_\_\_\_\_ than does **Solution C**; **Solution A** is also **hypertonic** to **Solution C**.



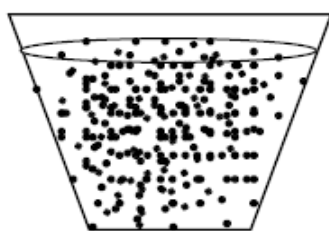
Cut along dotted line



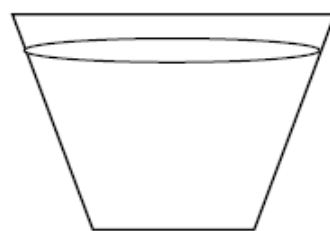
3. If you wanted to make **Solution A** isotonic to **Solution B**, you could add **water** to Solution \_\_\_\_ or you could add **solute** to Solution \_\_\_\_ . If you took all three solutions, put them into a large container and mixed them thoroughly, then redistributed the solution among the three containers, **Solution A** would be \_\_\_\_\_ to **Solution B**. **Solution A** would also be \_\_\_\_\_ to **Solution C**, and **Solution C** would be \_\_\_\_\_ to **Solution B**.



**A**



**B**

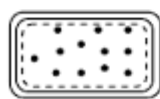


**C**

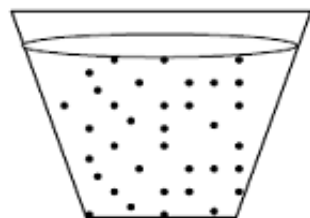
**Part III.** Below are represented a **plant cell** and an **animal cell**. Refer to the **key** at the top left of page one and fill in the blanks below. (If you find yourself counting solute dots, you're working **much** too hard!) Assume that the cell membranes are allow only water (not the solutes) to pass through.



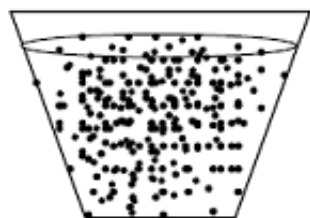
**animal cell**



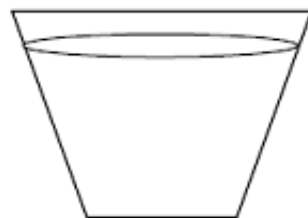
**plant cell**



**A**



**B**



**C**

1. Because the **cytoplasm**s of the plant and the animal cell have **equal** concentrations of solutes, we can say that their cytoplasm are \_\_\_\_\_ to each other. If we put both the plant and the animal cells into **Solution A**, we would expect **no change** in the cells, because **Solution A** is \_\_\_\_\_ to the cytoplasm of each cell.

2. Let's put both cells into **Solution B**. Because **Solution B** is **hypertonic** to the cytoplasm of the cells, we would expect **water** to \_\_\_\_\_ the cells through the process of \_\_\_\_\_. This would result in the cytoplasm of both cells shrinking.

3. Now we'll put both the plant and the animal cell into **Solution C**, which, because it contains **no solutes** at all, is \_\_\_\_\_ to the cytoplasm of both cells. \_\_\_\_\_ will enter both cells through **osmosis**. The **animal cell** is likely to \_\_\_\_\_, unfortunately. The **plant cell**, however, is protected from this because of the presence of its \_\_\_\_\_, which is lacking in the animal cell.